Concepts and Challenges (PS)
Unit 14 Chapter 4

Name		
Period	Student number	
Date		

## What is a polyatomic ion?

BLUE QUESTIONS. Copy and answer the blue questions in the space below.  1
2
3
WHAT YOU LEARNED. Copy each sentence in the space below.  1.
2.
SCIENCE WORD. Copy each term and its definition in the space below.  1.
2.
ANSWER THESE. Copy each sentence and fill in the blank with the correct answer.  1.
2.
3.
NOW TRY THIS. Match the formulas of the ions in column A by drawing connecting lines to their names in column

n B.

SO <sub>4</sub> -2	AMMONIUM
$NO_3^{-1}$	SULFATE
$NO_{3}^{-1}$ $NH_{4}^{+1}$	CARBONATE
$CO_{2}^{-2}$	NITRATE
$PO_4^{-2}$	PHOSPHATE

**FINDING OUT MORE.** Complete the following chart.

FORMULA	CHEMICAL NAME	COMMON NAME
NaOH		
Mg(OH) <sub>2</sub>		
Ca(OH) <sub>2</sub>		
NH <sub>4</sub> OH		